

The Synthesis of 2'-Deoxyadenosine via Stereospecific  
Coupling Reaction

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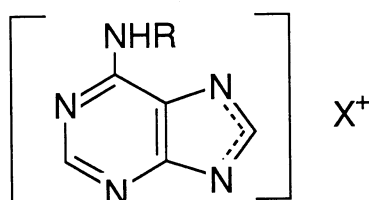
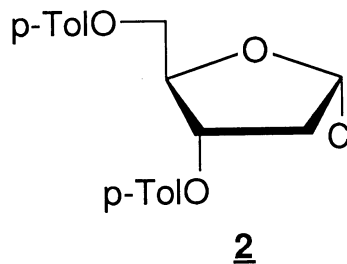
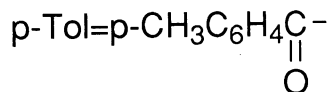
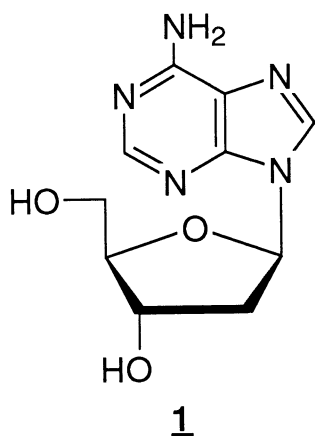
The coupling reaction of the sodium salt of adenine, which could be easily prepared by deprotonation with sodium hydroxide or sodium methoxide, with 1- $\alpha$ -chloro-2-deoxyribose derivative proceeded in a good stereospecific manner in acetone as a solvent to give the  $\beta$ -anomer of the corresponding acylated adenosine.

Much attention has recently been paid to 2'-deoxyadenosine (**1**) as a material for genetic engineering and as a source of many pharmacologically potent materials. For example, **1** is easily converted to 2',3'-dideoxyadenosine,<sup>1)</sup> which shows a significant antiviral activity against HTLV-III which causes Acquired Immune Deficiency Syndrome (AIDS).<sup>2)</sup> Because of no practical method for the chemical preparation, **1** is supplied in limited quantities only by separation from the hydrolyzed products of DNA of natural source. On a laboratory scale, the following two methods of the chemical preparation of **1** are known: One is the deoxygenation of adenosine,<sup>3)</sup> and the other is coupling reactions between adenine derivatives and 2-deoxyribose derivatives.<sup>4)</sup> Since it is difficult to differentiate the two secondary hydroxyl groups of adenosine, the former method is not useful for the practical purpose. In general, coupling reactions between 2-deoxyribose derivatives and nucleic bases with catalysts, especially Lewis acids, proceed in a non-stereospecific manner to give anomeric mixtures. In recent years, the coupling reactions between activated nucleic bases and 1- $\alpha$ -chloro-2-deoxyribose derivatives, for example 1- $\alpha$ -chloro-2-deoxy-3,5-di-O-(p-toluoxy)-D-ribose (**2**) which could be obtained as the pure form, were reported to proceed efficiently in S<sub>N</sub>2 mode at the C-1 position in the sugar to give only  $\beta$ -anomers. As activated nucleic bases, silylated uracil derivatives<sup>5)</sup> and the sodium salts of heterocyclic compounds, which are prepared *in situ*,<sup>6)</sup> have been reported. Especially, Robins et al. have reported the preparative method of 2'-deoxyadenosine derivatives by the coupling reaction between **2** and the sodium salts, prepared by *in situ* deprotonation of 6-chloropurine derivatives with NaH followed by the amination.<sup>6a)</sup> However, this method is not convenient for the preparation of 2'-deoxyadenosine **1**, because 6-chloropurine derivatives are rather expensive and the amination step requires somewhat higher pressure. In order to accomplish the

synthesis of **1** more efficiently, the coupling reaction between **2** and the salts of adenine derivatives was examined under various conditions.

At first, the coupling reaction between adenine and chlorosugar **2** was carried out under the conditions reported by Robins et al. (*in situ* deprotonation by NaH, acetonitrile as a solvent). However, the desired N-9 coupling products were obtained as the anomeric mixture in 35% yield. Since the H-9 proton of adenine is known to be sufficiently acidic ( $pK_a=9.80$ ) for deprotonation by NaOH or NaOMe, we tried to isolate the Na salt of adenine (**3a**). Addition of one equivalent of NaOH or NaOMe to a suspension of adenine in methanol resulted in disappearance of the solid. After removal of the solvent, white solid **3a**, which was stable to air and moisture, was obtained and used for the coupling reactions. The results are summarized in Table 1.

As the result of solvent effects (entries 2-10), the good anomeric specificities were achieved in moderate polar solvents, for instance, acetone and 1,2-dimethoxyethane (entries 8 and 9). These results were reasonable in terms of the following facts; the chlorosugar **2** was easily anomerized in polar solvents which was reported by Walker et al.,<sup>5a)</sup> and the Na salt **3a** was not soluble in non-polar organic solvents. However, the yields of N-9 coupling products were not so high because of low solubility of Na salt **3a**, therefore, we intended to raise the solubility. Benzoylation of the amino function made no remarkable change in yield (entries 11-15). The stereospecificities were lowered by the change of the counter cations of the salts (entries 16 and 17) or the addition of phase transfer catalysts (entries 1



**3a**: R=H ; X=Na  
**3b**: R=Bz; X=Na  
**3c**: R=H ; X=K  
**3d**: R=H ; X=Li

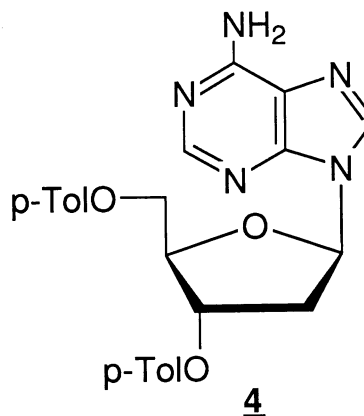


Table 1. The coupling reactions between chlorosugar **2** and salts of adenine derivatives **3a**)

| Entry            | Base      | Solvent                          | Yield/% <sup>b)</sup><br>( $\alpha$ + $\beta$ ) | Stereospecificity <sup>b)</sup><br>( $\alpha$ : $\beta$ ) |
|------------------|-----------|----------------------------------|---|---|
| 1 <sup>c)</sup>  | <b>3a</b> | CH <sub>3</sub> CN               | 35  | 16 : 84   |
| 2                | <b>3a</b> | CHCl <sub>3</sub>                | 14  | 36 : 64   |
| 3                | <b>3a</b> | CH <sub>2</sub> Cl <sub>2</sub>  | 24  | 15 : 85   |
| 4                | <b>3a</b> | PhH                              | 14  | 64 : 36   |
| 5                | <b>3a</b> | Et <sub>2</sub> O                | 17  | 16 : 84   |
| 6                | <b>3a</b> | THF                              | 23  | 6 : 94  |
| 7                | <b>3a</b> | EtOAc                            | 20  | 5 : 95  |
| 8                | <b>3a</b> | DME                              | 40  | 8 : 92  |
| 9                | <b>3a</b> | acetone                          | 50  | 17 : 83   |
| 10               | <b>3a</b> | DMF <sup>d)</sup>                | 35  | 41 : 59   |
| 11               | <b>3b</b> | CH <sub>3</sub> CN <sup>d)</sup> | 37  | 21 : 79   |
| 12               | <b>3b</b> | CH <sub>2</sub> Cl <sub>2</sub>  | 38  | 18 : 82   |
| 13               | <b>3b</b> | EtOAc                            | 36  | 4 : 96  |
| 14               | <b>3b</b> | DME                              | 46  | 5 : 95  |
| 15               | <b>3b</b> | acetone                          | 42  | 10 : 90   |
| 16               | <b>3c</b> | CH <sub>3</sub> CN <sup>d)</sup> | 18  | 57 : 43   |
| 17               | <b>3d</b> | CH <sub>3</sub> CN <sup>d)</sup> | 17  | 71 : 29   |
| 18 <sup>e)</sup> | <b>3a</b> | CH <sub>2</sub> Cl <sub>2</sub>  | 48  | 36 : 64   |
| 19 <sup>f)</sup> | <b>3a</b> | CH <sub>2</sub> Cl <sub>2</sub>  | 30  | 59 : 41   |
| 20 <sup>g)</sup> | <b>3a</b> | CH <sub>2</sub> Cl <sub>2</sub>  | 51  | 56 : 44   |

a) Coupling reactions were carried out under the following conditions; 0.25 mmol scale, sugar:salt=1:2, in 10 ml solvent, room temperature, for about 24 h.

b) Yields and stereospecificities were determined by HPLC (UV detection, 260 nm; compared to uracil as an internal standard) after deprotection by concentrated aqueous ammonia in methanol.

c) Coupling reaction was carried out under Robins' conditions. The isolated yield of **4** and its  $\alpha$  anomer was described. Stereospecificity was determined by <sup>1</sup>H-NMR.

d) In 2 ml solvent.

e) A catalytic amount of 18-crown-6 ether was added.

f) A catalytic amount of Bu<sub>4</sub>NHSO<sub>4</sub> was added.

g) A catalytic amount of BnEt<sub>3</sub>NCl was added.

8-20). The best result was obtained when the coupling reaction was carried out between **2** and Na salt **3a** in acetone.

A typical procedure is as follows. To a suspension of Na salt **3a** (0.40 g, 2.5 mmol) in 25 ml of acetone (dried over molecular sieves 3A) was added 1-chloro-2-deoxy-3,5-di-O-(p-toluoyl)- $\alpha$ -D-erythro-pentofuranose (**2**) (0.49 g, 1.3 mmol). The reaction mixture was stirred at room temperature for 19 hours under anhydrous conditions, and poured into brine, and extracted with CH<sub>2</sub>Cl<sub>2</sub>. After drying over anhydrous MgSO<sub>4</sub>, the organic layer was concentrated, and the residue was purified by

column chromatography (silica gel, CH<sub>2</sub>Cl<sub>2</sub>:acetone=50:50, v/v) to give 0.34 g of the anomeric mixture (yield 56%,  $\alpha$ : $\beta$ =20:80; determined by <sup>1</sup>H-NMR). Pure  $\beta$ -anomer **4** (0.29 g) was obtained in 43% yield by recrystallization from ethyl acetate. <sup>1</sup>H-NMR suggested that the crystals contained 1 molecule of ethyl acetate per 2 molecules of **4**.

In conclusion, we developed a simple procedure of the coupling reaction utilizing the sodium salt of adenine for the chemical preparation of 2'-deoxyadenosine. In moderate polar solvents such as acetone or 1,2-dimethoxyethane stereospecificities are rather good but yields are moderate. For increasing the yield further investigations are now under way.

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